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1-(4-Chloro-3-fluorophenyl)-2-[(3-phenylisoquinolin-1-yl)sulfanyl]ethanone

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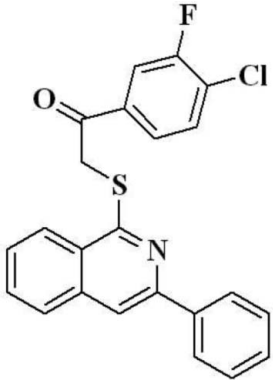
Key indicators: single-crystal X-ray study; T = 290 K; mean σ(C–C) = 0.004 Å; R factor = 0.056; wR factor = 0.128; data-to-parameter ratio = 14.2.

In the title compound, $C_{23}H_{15}ClFNOS$, the isoquinoline system and the 4-chloro-3-fluorophenyl ring are aligned at 80.4 (1)°. The dihedral angle between the isoquinoline system and the pendant (unsubstituted) phenyl ring is 19.91 (1)°.

Related literature

For related structures, see: Hathwar *et al.* (2008); Manivel *et al.* (2009*a*,*b*).



Experimental

Crystal data

$C_{23}H_{15}ClFNOS$
M_r = 407.87
Orthorhombic, *Pbca*
a = 16.9008 (11) Å
b = 9.8036 (7) Å
c = 23.3226 (16) Å
V = 3864.3 (5) Å^3
Z = 8
Mo $K\alpha$ radiation
μ = 0.33 mm^{-1}
T = 290 (2) K
0.24 × 0.18 × 0.11 mm

Data collection

Bruker SMART CCD area-detector diffractometer
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
T_{min} = 0.925, T_{max} = 0.965
27428 measured reflections
3595 independent reflections
2424 reflections with $I > 2\sigma(I)$
R_{int} = 0.063

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.056
$wR(F^2)$ = 0.128
S = 1.04
3595 reflections
253 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.32 e Å^{-3}
$\Delta\rho_{min}$ = −0.19 e Å^{-3}

Data collection: *SMART* (Bruker, 2004); cell refinement: *SAINT* (Bruker, 2004); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia,1997) and *CAMERON* (Watkin *et al.*, 1993); software used to prepare material for publication: *PLATON* (Spek, 2003).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: NG2534).

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supplementary materials

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1-(4-Chloro-3-fluorophenyl)-2-[(3-phenylisoquinolin-1-yl)sulfanyl]ethanone

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Comment

In compound (I), the S atom also located in the plane. The F atom deviates by 0.014 A from mean plane of phenyl ring containing F and Cl atoms. In this ring F– C and Cl—C bond distances are 1.348 (4) A, 1.727 (3) A, respectively. The orientation of isoquinoline ring system with respect to the another phenyl ring is given by the torsion angles for N1—C2—C10—C15 and C3—C2—C10—C11 are respectively -160.1 (2)°, -163.1 (3)° similarly for C16—S1—C1—N1 and C16—S1—C1—C8 are respectively -0.8 (2)° and 179.56 (19)° (Table 1).

Experimental

3-Phenylisoquinoline-1-thiol and 2-bromo-1-(3-fluoro-4-chlorophenyl)ethanone were mixed in the ratio 1:1.05 equivalents with ethanol in a round bottom flask. Then it was heated under nitrogen atmosphere on an oil bath at 323 K. After 2 h, the products were filtered and dissolved in chloroform. Further, it was washed with water, dried and concentrated. The single-crystal for X-ray structue anlaysis was obtained from ether solution by slow evaporation.

Refinement

All the H atoms in (I) were positioned geometrically and refined using a riding model with C—H bond lenghts of 0.93 Å and 0.97 Å for aromatic and for methylene H atoms respectively and U_{iso}(H) = 1.2U_{eq}(C) for all carbon bound H atoms.

Figures

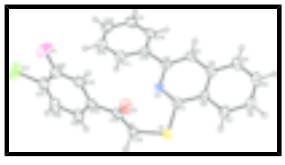

Fig. 1. *ORTEP* diagram of the asymmetric unit of (I) with 50% probability displacement ellipsoids.

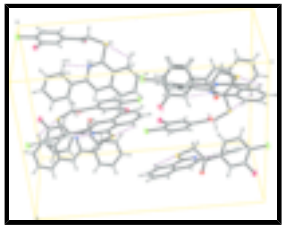

Fig. 2. The crystal packing diagram of (I).The dotted lines indicate intermolecular C—H···O hydrogen bonds. All H atoms have been omitted for clarity.

1-(4-Chloro-3-fluorophenyl)-2-[(3-phenylisoquinolin-1-yl)sulfanyl]ethanone

Crystal data

$C_{23}H_{15}ClFNOS$ F_{000} = 1680

M_r = 407.87	D_x = 1.402 Mg m^{-3}
Orthorhombic, *Pbca*	Mo *K*α radiation λ = 0.71073 Å
Hall symbol: -P 2ac 2ab	Cell parameters from 3595 reflections
a = 16.9008 (11) Å	θ = 1.8–25.5º
b = 9.8036 (7) Å	μ = 0.33 mm^{-1}
c = 23.3226 (16) Å	*T* = 290 (2) K
V = 3864.3 (5) Å^3	Block, colourless
Z = 8	0.24 × 0.18 × 0.11 mm

Data collection

Bruker SMART CCD area-detector diffractometer	3595 independent reflections
Radiation source: fine-focus sealed tube	2424 reflections with $I > 2\sigma(I)$
Monochromator: graphite	R_{int} = 0.063
T = 293(2) K	θ_{max} = 25.5º
φ and ω scans	θ_{min} = 1.8º
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	*h* = −18→20
T_{min} = 0.925, T_{max} = 0.965	*k* = −11→11
27428 measured reflections	*l* = −28→28

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)]$ = 0.056	H-atom parameters constrained
$wR(F^2)$ = 0.128	$w = 1/[\sigma^2(F_o^2) + (0.0566P)^2 + 1.1665P]$ where $P = (F_o^2 + 2F_c^2)/3$
S = 1.04	$(\Delta/\sigma)_{max}$ = 0.001
3595 reflections	$\Delta\rho_{max}$ = 0.32 e Å^{-3}
253 parameters	$\Delta\rho_{min}$ = −0.19 e Å^{-3}
Primary atom site location: structure-invariant direct methods	Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}

O1	0.07142 (12)	0.8183 (2)	0.28005 (9)	0.0633 (6)
F1	0.09785 (13)	0.8393 (2)	0.06905 (9)	0.1057 (7)
S1	0.03138 (4)	0.61721 (7)	0.37120 (3)	0.0495 (2)
Cl1	−0.02952 (8)	0.70329 (14)	0.00961 (4)	0.1221 (5)
N1	0.12270 (12)	0.4853 (2)	0.29715 (9)	0.0399 (5)
C1	0.11497 (15)	0.5201 (2)	0.35092 (11)	0.0393 (6)
C2	0.18572 (14)	0.4054 (2)	0.28104 (11)	0.0405 (6)
C3	0.24276 (16)	0.3684 (3)	0.31906 (11)	0.0474 (7)
H3	0.2861	0.3180	0.3066	0.057*
C4	0.29398 (18)	0.3701 (3)	0.41884 (13)	0.0593 (8)
H4	0.3384	0.3205	0.4079	0.071*
C5	0.2845 (2)	0.4076 (3)	0.47468 (14)	0.0688 (9)
H5	0.3225	0.3827	0.5016	0.083*
C6	0.2188 (2)	0.4828 (3)	0.49197 (13)	0.0643 (9)
H6	0.2135	0.5085	0.5302	0.077*
C7	0.16226 (18)	0.5189 (3)	0.45299 (12)	0.0538 (7)
H7	0.1180	0.5676	0.4650	0.065*
C8	0.17027 (16)	0.4832 (2)	0.39492 (11)	0.0422 (6)
C9	0.23657 (16)	0.4063 (3)	0.37753 (11)	0.0451 (6)
C10	0.18550 (14)	0.3638 (2)	0.21970 (11)	0.0419 (6)
C11	0.13897 (17)	0.4309 (3)	0.18011 (12)	0.0515 (7)
H11	0.1085	0.5046	0.1921	0.062*
C12	0.13641 (19)	0.3920 (3)	0.12340 (12)	0.0597 (8)
H12	0.1046	0.4392	0.0976	0.072*
C13	0.18110 (19)	0.2828 (3)	0.10499 (13)	0.0641 (9)
H13	0.1805	0.2569	0.0666	0.077*
C14	0.22641 (19)	0.2130 (4)	0.14375 (14)	0.0715 (10)
H14	0.2556	0.1379	0.1317	0.086*
C15	0.22938 (17)	0.2524 (3)	0.20044 (13)	0.0598 (8)
H15	0.2609	0.2042	0.2261	0.072*
C16	−0.01524 (15)	0.6351 (3)	0.30300 (11)	0.0439 (6)
H16A	−0.0177	0.5461	0.2849	0.053*
H16B	−0.0692	0.6659	0.3089	0.053*
C17	0.02538 (15)	0.7327 (2)	0.26264 (12)	0.0426 (6)
C18	0.00791 (15)	0.7226 (2)	0.20013 (12)	0.0422 (6)
C19	0.05916 (17)	0.7874 (3)	0.16253 (13)	0.0520 (7)
H19	0.1020	0.8366	0.1766	0.062*
C20	0.0465 (2)	0.7788 (3)	0.10523 (15)	0.0647 (9)
C21	−0.0164 (2)	0.7098 (4)	0.08298 (14)	0.0679 (9)
C22	−0.0680 (2)	0.6464 (3)	0.11953 (15)	0.0714 (9)
H22	−0.1116	0.6000	0.1049	0.086*
C23	−0.05585 (18)	0.6510 (3)	0.17826 (13)	0.0572 (8)
H23	−0.0904	0.6062	0.2029	0.069*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0674 (14)	0.0532 (12)	0.0694 (14)	−0.0142 (11)	−0.0162 (11)	−0.0004 (10)

F1	0.1066 (17)	0.1363 (19)	0.0742 (14)	−0.0033 (15)	0.0267 (12)	0.0272 (13)
S1	0.0554 (5)	0.0517 (4)	0.0412 (4)	0.0142 (3)	0.0010 (3)	−0.0073 (3)
Cl1	0.1660 (12)	0.1504 (11)	0.0500 (6)	0.0166 (9)	−0.0195 (6)	0.0006 (6)
N1	0.0408 (13)	0.0394 (12)	0.0396 (12)	0.0018 (10)	0.0016 (10)	−0.0030 (9)
C1	0.0454 (16)	0.0322 (13)	0.0404 (15)	−0.0012 (11)	0.0021 (12)	−0.0003 (11)
C2	0.0372 (15)	0.0380 (14)	0.0462 (15)	−0.0010 (11)	0.0047 (12)	−0.0020 (12)
C3	0.0381 (15)	0.0473 (15)	0.0570 (17)	0.0062 (12)	0.0002 (14)	−0.0042 (14)
C4	0.0579 (19)	0.0559 (18)	0.064 (2)	0.0096 (15)	−0.0138 (16)	0.0034 (16)
C5	0.081 (2)	0.068 (2)	0.058 (2)	0.0080 (19)	−0.0261 (18)	0.0111 (17)
C6	0.090 (2)	0.0590 (19)	0.0443 (17)	0.0110 (18)	−0.0115 (17)	0.0026 (14)
C7	0.070 (2)	0.0462 (16)	0.0448 (17)	0.0078 (15)	−0.0052 (14)	0.0015 (13)
C8	0.0501 (16)	0.0346 (13)	0.0419 (15)	−0.0011 (12)	−0.0037 (12)	0.0025 (11)
C9	0.0481 (16)	0.0368 (14)	0.0505 (16)	−0.0013 (12)	−0.0068 (13)	0.0021 (12)
C10	0.0368 (15)	0.0425 (14)	0.0464 (16)	−0.0034 (12)	0.0067 (12)	−0.0055 (12)
C11	0.0665 (19)	0.0400 (15)	0.0479 (17)	0.0057 (14)	0.0020 (15)	−0.0024 (13)
C12	0.077 (2)	0.0564 (17)	0.0457 (17)	0.0053 (16)	−0.0019 (15)	0.0004 (14)
C13	0.065 (2)	0.081 (2)	0.0468 (18)	0.0032 (18)	0.0075 (16)	−0.0157 (16)
C14	0.058 (2)	0.091 (3)	0.065 (2)	0.0256 (19)	0.0001 (17)	−0.0299 (19)
C15	0.0466 (18)	0.074 (2)	0.0585 (19)	0.0217 (16)	−0.0017 (14)	−0.0164 (16)
C16	0.0429 (16)	0.0415 (15)	0.0472 (16)	0.0076 (12)	0.0010 (12)	−0.0031 (12)
C17	0.0383 (15)	0.0353 (14)	0.0543 (17)	0.0055 (12)	−0.0039 (13)	−0.0023 (12)
C18	0.0410 (15)	0.0341 (13)	0.0514 (17)	0.0041 (12)	−0.0020 (13)	0.0016 (12)
C19	0.0485 (17)	0.0474 (17)	0.060 (2)	0.0057 (13)	0.0024 (14)	0.0034 (14)
C20	0.069 (2)	0.069 (2)	0.056 (2)	0.0130 (18)	0.0129 (18)	0.0134 (16)
C21	0.087 (3)	0.072 (2)	0.0453 (18)	0.018 (2)	−0.0032 (18)	0.0037 (16)
C22	0.081 (2)	0.070 (2)	0.064 (2)	−0.0032 (18)	−0.0258 (19)	−0.0049 (17)
C23	0.0612 (19)	0.0512 (17)	0.059 (2)	−0.0033 (14)	−0.0071 (16)	0.0041 (14)

Geometric parameters (Å, °)

O1—C17	1.215 (3)	C10—C15	1.395 (3)
F1—C20	1.348 (4)	C11—C12	1.377 (4)
S1—C1	1.768 (3)	C11—H11	0.9300
S1—C16	1.784 (3)	C12—C13	1.378 (4)
Cl1—C21	1.727 (3)	C12—H12	0.9300
N1—C1	1.306 (3)	C13—C14	1.368 (4)
N1—C2	1.375 (3)	C13—H13	0.9300
C1—C8	1.434 (3)	C14—C15	1.378 (4)
C2—C3	1.359 (3)	C14—H14	0.9300
C2—C10	1.487 (3)	C15—H15	0.9300
C3—C9	1.417 (3)	C16—C17	1.508 (4)
C3—H3	0.9300	C16—H16A	0.9700
C4—C5	1.363 (4)	C16—H16B	0.9700
C4—C9	1.413 (4)	C17—C18	1.491 (4)
C4—H4	0.9300	C18—C23	1.383 (4)
C5—C6	1.392 (4)	C18—C19	1.387 (4)
C5—H5	0.9300	C19—C20	1.356 (4)
C6—C7	1.366 (4)	C19—H19	0.9300
C6—H6	0.9300	C20—C21	1.363 (5)

C7—C8	1.405 (4)	C21—C22	1.369 (5)
C7—H7	0.9300	C22—C23	1.386 (4)
C8—C9	1.410 (3)	C22—H22	0.9300
C10—C11	1.380 (4)	C23—H23	0.9300
C1—S1—C16	99.63 (12)	C13—C12—H12	120.1
C1—N1—C2	119.3 (2)	C14—C13—C12	119.3 (3)
N1—C1—C8	123.8 (2)	C14—C13—H13	120.3
N1—C1—S1	118.51 (19)	C12—C13—H13	120.3
C8—C1—S1	117.71 (19)	C13—C14—C15	120.9 (3)
C3—C2—N1	121.5 (2)	C13—C14—H14	119.5
C3—C2—C10	123.8 (2)	C15—C14—H14	119.5
N1—C2—C10	114.7 (2)	C14—C15—C10	120.6 (3)
C2—C3—C9	120.4 (2)	C14—C15—H15	119.7
C2—C3—H3	119.8	C10—C15—H15	119.7
C9—C3—H3	119.8	C17—C16—S1	114.70 (19)
C5—C4—C9	120.2 (3)	C17—C16—H16A	108.6
C5—C4—H4	119.9	S1—C16—H16A	108.6
C9—C4—H4	119.9	C17—C16—H16B	108.6
C4—C5—C6	120.9 (3)	S1—C16—H16B	108.6
C4—C5—H5	119.6	H16A—C16—H16B	107.6
C6—C5—H5	119.6	O1—C17—C18	120.0 (2)
C7—C6—C5	120.2 (3)	O1—C17—C16	121.5 (3)
C7—C6—H6	119.9	C18—C17—C16	118.6 (2)
C5—C6—H6	119.9	C23—C18—C19	119.1 (3)
C6—C7—C8	120.6 (3)	C23—C18—C17	123.3 (3)
C6—C7—H7	119.7	C19—C18—C17	117.7 (2)
C8—C7—H7	119.7	C20—C19—C18	119.7 (3)
C7—C8—C9	119.2 (2)	C20—C19—H19	120.1
C7—C8—C1	124.3 (2)	C18—C19—H19	120.1
C9—C8—C1	116.5 (2)	F1—C20—C19	119.2 (3)
C8—C9—C4	118.9 (3)	F1—C20—C21	118.8 (3)
C8—C9—C3	118.4 (2)	C19—C20—C21	122.0 (3)
C4—C9—C3	122.7 (3)	C20—C21—C22	119.0 (3)
C11—C10—C15	117.4 (2)	C20—C21—Cl1	119.7 (3)
C11—C10—C2	120.9 (2)	C22—C21—Cl1	121.2 (3)
C15—C10—C2	121.6 (2)	C21—C22—C23	120.4 (3)
C12—C11—C10	121.9 (3)	C21—C22—H22	119.8
C12—C11—H11	119.0	C23—C22—H22	119.8
C10—C11—H11	119.0	C18—C23—C22	119.8 (3)
C11—C12—C13	119.8 (3)	C18—C23—H23	120.1
C11—C12—H12	120.1	C22—C23—H23	120.1
C2—N1—C1—C8	2.0 (4)	C15—C10—C11—C12	−1.1 (4)
C2—N1—C1—S1	−177.53 (17)	C2—C10—C11—C12	−178.4 (3)
C16—S1—C1—N1	−0.8 (2)	C10—C11—C12—C13	0.2 (4)
C16—S1—C1—C8	179.56 (19)	C11—C12—C13—C14	1.2 (5)
C1—N1—C2—C3	−4.0 (4)	C12—C13—C14—C15	−1.6 (5)
C1—N1—C2—C10	175.8 (2)	C13—C14—C15—C10	0.6 (5)
N1—C2—C3—C9	3.1 (4)	C11—C10—C15—C14	0.7 (4)

C10—C2—C3—C9	−176.7 (2)	C2—C10—C15—C14	178.0 (3)
C9—C4—C5—C6	−0.5 (5)	C1—S1—C16—C17	−73.20 (19)
C4—C5—C6—C7	0.7 (5)	S1—C16—C17—O1	−19.3 (3)
C5—C6—C7—C8	−1.2 (4)	S1—C16—C17—C18	160.73 (18)
C6—C7—C8—C9	1.6 (4)	O1—C17—C18—C23	−164.7 (3)
C6—C7—C8—C1	−178.2 (3)	C16—C17—C18—C23	15.2 (4)
N1—C1—C8—C7	−179.4 (2)	O1—C17—C18—C19	16.0 (4)
S1—C1—C8—C7	0.1 (3)	C16—C17—C18—C19	−164.1 (2)
N1—C1—C8—C9	0.7 (4)	C23—C18—C19—C20	−0.6 (4)
S1—C1—C8—C9	−179.71 (18)	C17—C18—C19—C20	178.7 (2)
C7—C8—C9—C4	−1.5 (4)	C18—C19—C20—F1	−178.3 (2)
C1—C8—C9—C4	178.4 (2)	C18—C19—C20—C21	1.2 (5)
C7—C8—C9—C3	178.5 (2)	F1—C20—C21—C22	179.0 (3)
C1—C8—C9—C3	−1.6 (3)	C19—C20—C21—C22	−0.5 (5)
C5—C4—C9—C8	0.9 (4)	F1—C20—C21—Cl1	−0.8 (4)
C5—C4—C9—C3	−179.1 (3)	C19—C20—C21—Cl1	179.7 (2)
C2—C3—C9—C8	−0.2 (4)	C20—C21—C22—C23	−0.8 (5)
C2—C3—C9—C4	179.8 (3)	Cl1—C21—C22—C23	179.0 (2)
C3—C2—C10—C11	−163.1 (3)	C19—C18—C23—C22	−0.7 (4)
N1—C2—C10—C11	17.1 (3)	C17—C18—C23—C22	−180.0 (3)
C3—C2—C10—C15	19.7 (4)	C21—C22—C23—C18	1.4 (5)
N1—C2—C10—C15	−160.1 (2)		

Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
C7—H7···S1	0.93	2.68	3.076 (3)	107
C11—H11···N1	0.93	2.47	2.795 (4)	101

Fig. 1

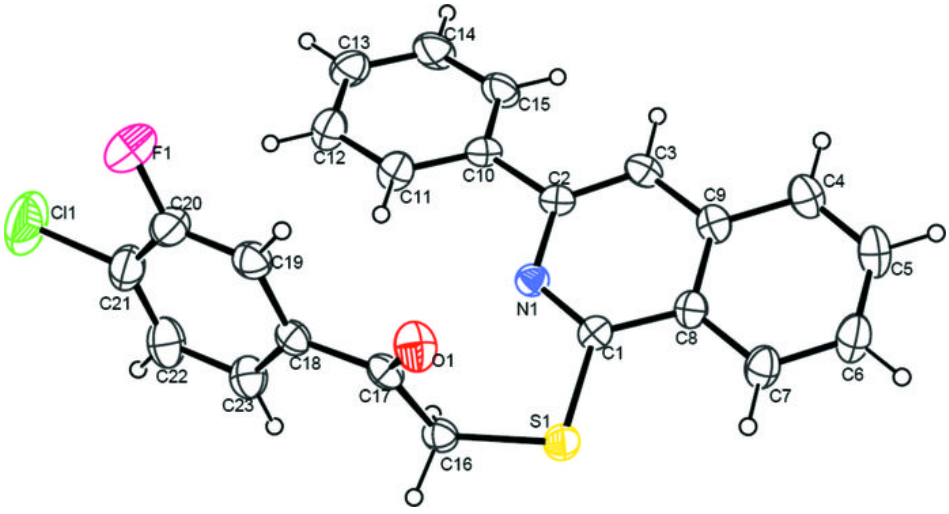
C13
C14
C15
F1
C12
C10
C2
C3
C4
C11
C9
Cl1
C20
C19
C5
C21
N1
C8
C1
C18
O1
C22
C6
C23
C17
C7
S1
C16

Fig. 2